

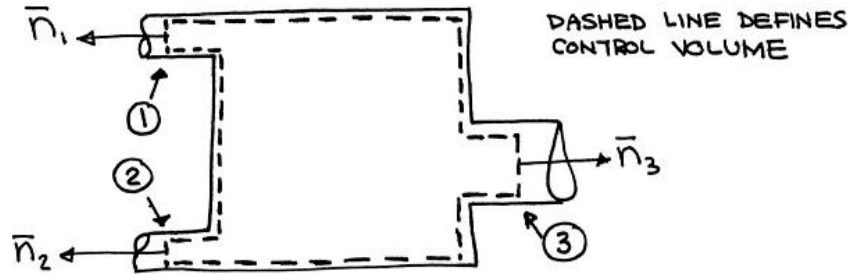
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1.061 / 1.61 Transport Processes in the Environment
Fall 2008

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Solution 1

A convenient control volume can be drawn around the interior volume of the tank, and extending into pipes 1 and 2 to positions of uniform concentration, i.e. $\frac{\partial C}{\partial n} = 0$ along pipe.



Now evaluate Eq.4 for this control volume.

(A)

$$\frac{\partial}{\partial t} \int_{CV} C dV = - \int_{CS} C \vec{V} \cdot \vec{n} dA + \int_{CS} D_n \frac{\partial C}{\partial n} dA \pm S$$

Because we assume steady state, $\frac{\partial}{\partial t} = 0$, the first term is zero. No source or sink is mentioned, \therefore set $S = 0$. We evaluate the two surface integrals, \int_{CS} , at the three indicated areas of flux. Note that we placed the surface 1, 2, 3 far enough into the pipes that $\frac{\partial C}{\partial n} = 0$ at each surface. \therefore there is no diffusive flux, $\int_{CS} D_n \frac{\partial C}{\partial n} dA = 0$.

(B) Evaluating $\int_{CS} C \vec{V} \cdot \vec{n} dA$ at each flux area,

$$0 = +u_1 A_1 C_1 + u_2 A_2 C_2 - u_3 A_3 C_3.$$

From conservation of fluid mass (continuity), we also have $u_1 A_1 + u_2 A_2 = u_3 A_3$ for incompressible flow.

(C) Using this to replace $u_3 A_3$ in (B) and solving for C_3 ,

$$C_3 = \frac{u_1 A_1 C_1 + u_2 A_2 C_2}{(u_1 A_1 + u_2 A_2)}$$

or,

$$C_3 = \frac{(20 \text{ cm/s})(10 \text{ cm}^2)(9 \text{ mg/l}) + (10 \text{ cm/s})(10 \text{ cm}^2)(0 \text{ mg/l})}{(20 \text{ cm/s})(10 \text{ cm}^2) + (10 \text{ cm/s})(10 \text{ cm}^2)}$$

$$C_3 = \frac{20}{30} * 9 \text{ mg/l} = 6 \text{ mg/l}$$

Solution 2

Apply the integral form of mass conservation to the control volume indicated by dashes.

$$\frac{\partial}{\partial t} \int_{CV} C dV = - \int_{CS} C \vec{V} \cdot \vec{n} dA + \int_{CS} D_n \frac{\partial C}{\partial n} dA + S$$

(A) \emptyset , b/c we assume steady state evaluate at surface sections 1,2 \emptyset , b/c we place surfaces 1 & 2 where $\frac{\partial C}{\partial n} = 0$ given

(B)

$$0 = u_1 C_1 A_1 - u_2 C_2 A_2 + S$$

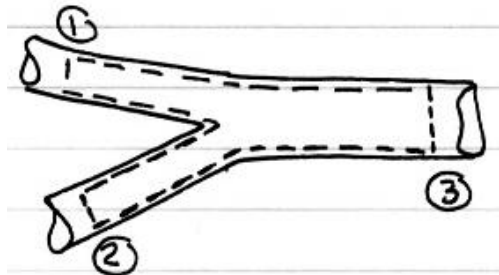
Note: From continuity, $u_2 = u_1$, because $A_2 = A_1$

(C)

$$C_2 = \frac{S}{u_2 A_2} = \frac{S}{u_1 A_1} = \frac{5 \text{ g/s}}{(10 \text{ cm/s})(10 \text{ cm}^2)} = 50 \text{ mg/cm}^3$$

Solution 3

Choose a control volume (dash) far enough away from juncture such that $\frac{\partial T}{\partial n} = 0$ at each flux surface.



(A)

$$\frac{\partial}{\partial t} \int_{CV} C dV = - \int_{CS} C \vec{V} \cdot \vec{n} dA + \int_{CS} D_n \frac{\partial C}{\partial n} dA \pm S$$

The concentration of heat energy is,

$$C [\text{Jm}^{-3}] = \rho c_p T$$

fluid density [kgm^{-3}]	specific heat [$\text{Jkg}^{-1}\text{K}^{-1}$]	temp [K]
$C_p = 4200 \text{ Jkg}^{-1}\text{K}^{-1}$ for water		

For simplicity, assume $\rho, c_p \neq f(T)$. If we assume steady state, then the first term in (A) is zero. Because we position surfaces 1, 2, 3 where $\frac{\partial T}{\partial n} = 0$, the diffusive flux term is zero. Because the pipes are insulated, $S = 0$. So, finally (A) becomes,

(B)

$$0 = \rho c_p T_1 u_1 A_1 + \rho c_p T_2 u_2 A_2 - \rho c_p T_3 u_3 A_3$$

Dropping ρc_p , and solving for T_3 ,

(C)

$$T_3 = \frac{u_1 A_1 T_1 + u_2 A_2 T_2}{u_3 A_3}$$

Note from statement $u_1 A_1 = u_2 A_2$. And from fluid mass conservation $(u_1 A_1 + u_2 A_2) = u_3 A_3$.

(D)

$$T_3 = \frac{u_1 A_1 (T_1 + T_2)}{2u_1 A_1} = \frac{1}{2} (T_1 + T_2)$$

$$\therefore T_3 = 15 \text{ deg } C \quad [288K]$$

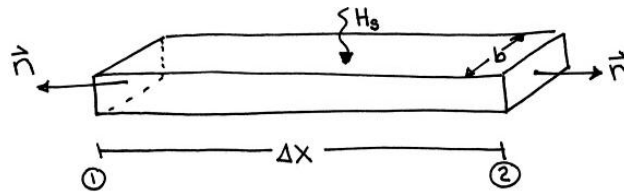
Solution 4

Control Volume Approach:

Select a short length of river, ΔX , and evaluate the control volume (integral) form of the conservation equation. For conservation of heat energy, replace $C = \rho c_p T$ in Eq.4.

(A)

$$\frac{\partial}{\partial t} \int_{CV} \rho c_p T dV = - \int_{CS} \rho c_p T \vec{V} \cdot \vec{n} dA + \int_{CS} D_n \frac{\partial C}{\partial n} dA + H_s \Delta x b$$



As the problem statement does not indicate any unsteadiness, we assume steady flow, i.e. $\frac{\partial}{\partial t} = 0$.

(B) Evaluating the flux terms in (A),

$$0 = -\rho c_p U b h (T_2 - T_1) + \rho c_p D b h \left(\left. \frac{\partial T}{\partial X} \right|_2 - \left. \frac{\partial T}{\partial X} \right|_1 \right) + H_3 \Delta x b$$

Using a Taylor expansion, assuming T is continuous in X ,

$$T_2 = T_1 + \frac{\partial T}{\partial X} \Delta X$$

$$\left. \frac{\partial T}{\partial X} \right|_2 = \left. \frac{\partial T}{\partial X} \right|_1 + \frac{\partial}{\partial X} \left(\frac{\partial T}{\partial X} \right) \Delta X$$

Plug these expansions in (B), and divide out the common terms, $\Delta x b$.

(C)

$$0 = \rho c_p h \left(-U \frac{\partial T}{\partial X} + D \frac{\partial^2 T}{\partial X^2} \right) + H_s$$

From which, one could solve for $\frac{\partial T}{\partial X}$.

It is useful to consider the relative importance of the advective and diffusive fluxes. Here, specifically the relative magnitudes of $U \frac{\partial T}{\partial X}$ and $D \frac{\partial^2 T}{\partial X^2}$. The scale of each term can be estimated from this system. Consider the control volume length, Δx , as an appropriate length-scale, then

$$U \frac{\partial T}{\partial X} \sim U \frac{\Delta T}{\Delta X}$$

$$D \frac{\partial^2 T}{\partial X^2} \sim D \frac{\Delta T}{\Delta X^2}$$

Where ΔT is the temperature change across ΔX . The relative magnitude of these terms is then,

$$\frac{\text{advective flux}}{\text{diffusive flux}} = \frac{U \frac{\Delta T}{\Delta X}}{D \frac{\Delta T}{\Delta X^2}} = \frac{U \Delta X}{D}$$

This dimensionless parameter is called the pecllet number. It is discussed in detail in Chapter 5.

If $\frac{U \Delta X}{D} \gg 1$, then advective fluxes dominate diffusive fluxes, and we can drop the term $D \frac{\partial^2 T}{\partial X^2} \ll U \frac{\partial T}{\partial X}$.

If $\frac{U \Delta X}{D} \ll 1$, diffusive fluxes $\left(D \frac{\partial^2 T}{\partial X^2} \right)$ are much larger than advective fluxes $\left(U \frac{\partial T}{\partial X} \right)$, and we can drop $U \frac{\partial T}{\partial X}$.

Since ΔX is not specifically defined, we ask, e.g., for what length-scale will advection dominate transport?

$$\frac{U \Delta X}{D} \gg 1 \text{ iff } \Delta X \gg \frac{D}{U} = \frac{0.1 \text{ m}^2 \text{ s}^{-1}}{1 \text{ m}^2 \text{ s}^{-1}} = 0.1 \text{ m}$$

\therefore over any length-scale, $\Delta X \gg 0.1 m$, we may neglect the impact of diffusion in (A) for this system. The problem asks for a description of $\frac{\partial T}{\partial X}$ along a river channel. In such a system, the length scales of interest are much larger than 10 cm, and are more like 100 m to km's. Therefore, for this system, we can safely drop the diffusive transport term. Then, (C) reduces to,

(D)

$$0 = -\rho c_p h U \frac{\partial T}{\partial X} + H_s$$

from which,

(E)

$$\frac{\partial T}{\partial X} = \frac{H_s}{\rho c_p h U} = \frac{Js^{-1}m^{-2}}{(kgm^{-3})(Jkg^{-1}K^{-1})(m)(ms^{-1})} = \frac{K}{m}$$

Using the stated parameters,

(F)

$$\frac{\partial T}{\partial X} = \frac{800 W m^{-2}}{(1000 kg/m^3)(4200 J/kgK)(1 m)(1 m/s)} = 2 \times 10^{-4} K/m = 0.2 C/km$$

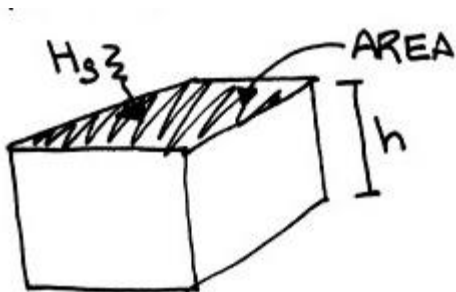
Differential Approach:

The conservation of mass equation can be applied to the transport of heat energy by noting the concentration of heat energy, $C[J/m^3] = \rho c_p T$. Then, the differential form of the conservation equation is, for incompressible flow,

(G)

$$\begin{aligned} & \frac{\partial}{\partial t} (\rho c_p T) + U \frac{\partial}{\partial x} (\rho c_p T) + V \frac{\partial}{\partial y} (\rho c_p T) + W \frac{\partial}{\partial z} (\rho c_p T) \\ &= \frac{\partial}{\partial x} D_x \frac{\partial}{\partial x} (\rho c_p T) + \frac{\partial}{\partial y} D_y \frac{\partial}{\partial y} (\rho c_p T) + \frac{\partial}{\partial z} D_z \frac{\partial}{\partial z} (\rho c_p T) \pm S \end{aligned}$$

- If we neglect $\rho = f(T)$, then $\rho \neq f(x, y, z, t)$.
- The problem statement gives us $V = W = 0$, and isotropic, homogeneous $D = D_x = D_y = D_z \neq f(x, y, z)$.
- If we assume the system is uniform (well-mixed) in y and z, then $T \neq f(y, z)$.
- The source term is given as a surface flux, $H_s = [Js^{-1}m^{-2}]$. Since the equation deals in volume concentration, we must divide by depth to put the source term in consistent units.



$$H_s = \frac{J/m^2}{s} = \frac{\text{energy per surface area}}{s}$$

$$\frac{H_s}{h} = \frac{J/m^3}{s} = \frac{\text{energy per volume}}{s}$$

Applying the above points, (G) reduces to

(H)

$$\rho c_p \left[\frac{\partial T}{\partial t} + u \frac{\partial T}{\partial X} \right] = D \frac{\partial^2 T}{\partial X^2} + \frac{H_s}{h}$$

For typical length scales of interest along a river channel, $\Delta X \sim 100m$ to km's, it is easy to show that the diffusion term, $D \frac{\partial^2 T}{\partial X^2}$, is small compared to $U \frac{\partial T}{\partial X}$, the advection term. Thus we will drop $D \frac{\partial^2 T}{\partial X^2} \ll U \frac{\partial T}{\partial X}$.

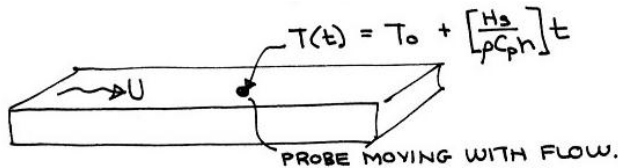
See scaling arguments given above.

Finally, note that the bracketed term in (H) is the total derivative.

(I)

$$\frac{DT}{Dt} = \frac{\partial T}{\partial t} + U \frac{\partial T}{\partial X} = \frac{H_s}{\rho c_p h} = \left[\frac{^\circ C}{T} \right]$$

This equation may be read in the Lagrangian context as, following a particular fluid particle, we would observe its temperature to increase at the rate $\left[\frac{H_s}{\rho c_p h} \right]^\circ C/s$.

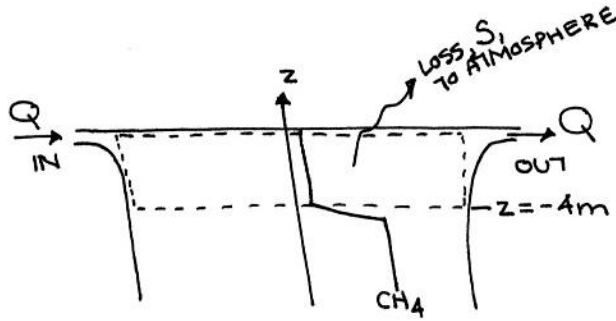


If the flow/thermal conditions are steady, $\frac{\partial T}{\partial t} = 0$, then (I) also provides a simple description of spatial gradient.

(J)

$$\frac{\partial T}{\partial X} = \frac{H_s}{U \rho c_p h} = \left[\frac{^\circ C}{L} \right]$$

Solution 5



Define the epilimnion as the control volume of interest. The conservation of mass for this system, assuming steady state, is

(A)

$$0 = (QC)_{IN} - (QC)_{OUT} - D_Z \frac{\partial C}{\partial Z} A - S$$

given $C_{IN} = 0$. Assume epilimnion is well-mixed. $\therefore C_{OUT} = C_{EPI} = C(z = -4m)$. Discretize concentration gradient across thermocline,

$$\left. \frac{\partial C}{\partial Z} \right|_{z=-4} \approx \frac{[C(z = -4) - C(z = -5)]}{[-4 - -5]}$$

Then solve (A) for S,

(B)

$$\begin{aligned} S &= -QC_{EPI} - D_Z A \frac{C_{EPI} - C_{HYPO}}{1 \text{ m}} \\ &= -(1 \text{ m}^3/\text{s})(0.4 \text{ mg/l}) - (2 \times 10^{-6} \text{ m}^2 \text{ s}^{-1})(10^6 \text{ m}^2) \left[\frac{0.4 - 4 \text{ mg/l}}{1 \text{ m}} \right] \\ &= 6.8 \frac{\text{m}^3}{\text{s}} \frac{\text{mg}}{\text{l}} \frac{1000 \text{ l}}{\text{m}^3} = 6.8 \text{ g/s} \end{aligned}$$

Solution 6

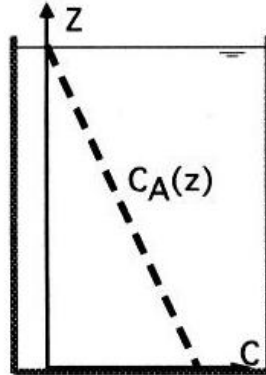
From Fick's Law, $q_z = -DA \frac{\partial C}{\partial Z}$.

(1) At $z = 0$ and $z = H$, $\frac{\partial C}{\partial Z} < 0$, $\therefore q_z > 0$, which indicates fluxes is upward at both boundaries.

(2) At steady state, $\frac{\partial C}{\partial t} = 0$. From 1-D conservation equation,

$$\frac{\partial C}{\partial t} = D_z \frac{\partial^2 C}{\partial Z^2}$$

\therefore at steady state, $\frac{\partial^2 C}{\partial Z^2} = 0$. $\therefore \frac{\partial C}{\partial Z} = \text{constant}$. \therefore linear profile indicates this system is at steady state.

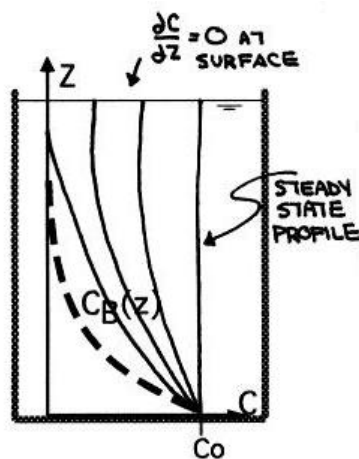


(1) $\frac{\partial C}{\partial Z} < 0$ at $z = 0$. $\therefore q_z > 0$, positive, upward. No flux at $z = H$.

(2) System is not at steady state.

$$\frac{\partial}{\partial Z} \left(\frac{\partial C}{\partial Z} \right) \neq 0, \therefore \text{from conservation EQ, } \frac{\partial C}{\partial Z} \neq 0$$

Progression of evolving profiles shown. Note, because chemical is not volatile, surface ($z = H$) is a no flux boundary. $\therefore \frac{\partial C}{\partial Z} \Big|_{z=H} = 0$ so that $q_z(z = H) = 0!$



Solution 7

(a) The general equation (14b) is

$$\begin{array}{cccccccc} \frac{\partial C}{\partial t} & + u \frac{\partial C}{\partial x} & + v \frac{\partial C}{\partial y} & + w \frac{\partial C}{\partial z} & = D \left[\frac{\partial^2 C}{\partial x^2} & + \frac{\partial^2 C}{\partial y^2} & + \frac{\partial^2 C}{\partial z^2} \right] & \pm S \\ (1) & (2) & (3) & (4) & (5) & (6) & (7) & (8) \end{array}$$

We can eliminate several terms in this equation:

- (1) because we are looking at steady-state conditions,
- (2), (3), (5) and (6) due to the negligible horizontal gradients of DO concentration and
- (4) due to the insignificant vertical flows.

Thus, the equation reduces to

$$0 = D \left[\frac{\partial^2 C}{\partial z^2} \right] - S$$

noting that phytoplankton are a sink of dissolved oxygen. For first-order removal, $S = kC$, so our governing equation is:

$$D \left[\frac{\partial^2 C}{\partial z^2} \right] - kC = 0$$

(b) The general solution to this governing equation is:

$$C(z) = A \exp \left(z \sqrt{\frac{k}{D}} \right) + B \exp \left(-z \sqrt{\frac{k}{D}} \right)$$

where A and B are constants. We use our boundary conditions to determine A and B.

- $C|_{z=0} = 8 \Rightarrow A + B = 8$
- No flux at $z = -10 \text{ m} \Rightarrow \left. \frac{\partial C}{\partial z} \right|_{z=-10} = 0$.

Therefore,

$$\begin{aligned} A \sqrt{\frac{k}{D}} \exp \left((-10 \text{ m}) \sqrt{\frac{k}{D}} \right) - B \sqrt{\frac{k}{D}} \exp \left(-(-10 \text{ m}) \sqrt{\frac{k}{D}} \right) &= 0 \\ \Rightarrow A \exp \left((-10 \text{ m}) \sqrt{\frac{k}{D}} \right) &= B \exp \left((10 \text{ m}) \sqrt{\frac{k}{D}} \right). \end{aligned}$$

Using our values for k and D,

$$\sqrt{\frac{k}{D}} = \sqrt{\frac{0.05 \text{ day}^{-1}}{0.1 \text{ cm}^2 \text{ s}^{-1}}} = \sqrt{\frac{5.787 \times 10^{-7} \text{ s}^{-1}}{10^{-5} \text{ m}^2 \text{ s}^{-1}}} = 0.240/\text{m}$$

Therefore, from our second boundary condition,

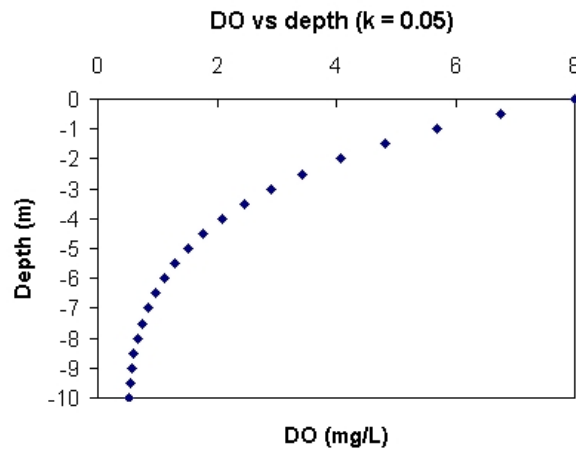
$$A \exp(-2.4) = B \exp(2.4) \Rightarrow 0.091A = 11.0B \Rightarrow A = 121.5B$$

Solving the boundary condition equations simultaneously gives $A = 7.935$, $B = 0.065$.

Therefore,

$$C(z) = 7.935\exp(0.240z) + 0.065\exp(-0.240z) \quad [\text{for } z \text{ in meters}]$$

- (c) As we have only a DO sink in the water column, the lowest concentration of DO will be found at the point furthest from the surface, which is maintained at a constant concentration. Thus, we expect the DO concentration to be lowest at $z = -10$ m. This is confirmed by looking at the plot of the vertical profile of DO in the water column (note that we have satisfied both boundary conditions).



- (d) As we can see from the above plot, $C_{min} = 1.4 \text{ mg/L}$. However, if $k = 0.1 \text{ day}^{-1}$, repetition of the above calculation yields a minimum concentration of approximately 0.5 mg/L , meaning the the health of Lake Monger is under threat.